

NEW DEPSIDONES FROM *LOBARIA OREGANA*SACHIKO SHIMADA (née MIYOSHI)\*, TAMOTSU SAITOH†,  
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**Key Word Index**—*Lobaria oregana*; lichen metabolites; methylstictic acid; crypto-stictic acid; cryptostictinolide.**Abstract**—From the lichen, *Lobaria oregana*, two new minor metabolites, methylstictic acid (**4**) and cryptostictic acid (**5**) were isolated. The structures of **4** and **5** were elucidated by  $^1\text{H}$  NMR and MS analysis.

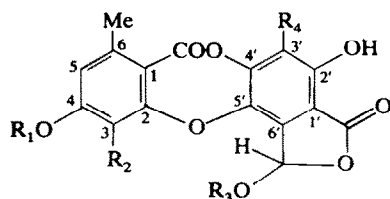
The metabolites of the lichen, *Lobaria oregana* (Tuck.) Müll. Ar. were studied earlier by Asahina *et al.* [1] and Culberson [2] to report the occurrence of (+)-usnic acid, norstictic acid (**1**), stictic acid (**2**) and constictic acid (**3**). The present study concerns minor constituents of *L. oregana* collected in British Columbia, Canada.

The lichen material was first extracted with  $\text{C}_6\text{H}_6$  to isolate (+)-usnic acid and ergosterol peroxide, then extracted at room temperature successively with  $\text{CHCl}_3$  and  $\text{Me}_2\text{CO}$ .

The  $\text{Me}_2\text{CO}$  extracts were chromatographed over a 0.5 N oxalic acid-impregnated Si gel column using  $\text{C}_6\text{H}_6$ - $\text{Me}_2\text{CO}$  as the solvent. TLC of the eluate developed with  $\text{C}_6\text{H}_6$ -dioxan-HOAc (90:25:4) gave more than 7 spots among which norstictic acid (**1**), stictic acid (**2**) and constictic acid (**3**) were identified, while the occurrence of two new compounds named methylstictic acid (**4**) and cryptostictic acid (**5**) was observed along with some artifacts.

lar to stictic acid (**2**), and also showed very similar absorptions of UV and IR to those of **2**. The  $^1\text{H}$  NMR in  $\text{DMSO}-d_6$  of **4** gave signals for two methyls ( $\delta$  2.48, 2.22), one each for methoxyl ( $\delta$  3.91) and aldehyde ( $\delta$  10.39) attached to aromatic rings at almost the same chemical shifts given by **2**, whereas an additional methoxyl signal at  $\delta$  3.44 was given by **4** in place of a signal for a lactol OH at  $\delta$  8.1 observed for **2**. The MS of **4** showed a  $\text{M}^+$  at  $m/e$  400 and a peak  $m/e$  368 ( $\text{M}^+ - \text{MeOH}$ , while **2** gave  $\text{M}^+$   $m/e$  386 and  $\text{M}^+ - \text{H}_2\text{O}$  at  $m/e$  368 [3] (Scheme 1). On these results, methylstictic acid has been represented by the structure (**4**).

Cryptostictic acid (**5**),  $\text{C}_{19}\text{H}_{16}\text{O}_9$ , colourless needles, mp  $242\sim 244^\circ$  (decomp.) showed a very similar  $^1\text{H}$  NMR spectrum to that of stictic acid (**2**), the only difference being the absence of the signal for CHO and the appearance of one for  $\text{CH}_2\text{OH}$  at  $\delta$  4.6 (1H *br. d*  $J = 10$  Hz) and  $\delta$  4.72 (1H *br. d*  $J = 10$  Hz). On acetylation **5** afforded a triacetate to confirm the presence of a  $\text{CH}_2\text{OH}$  group. The location of the  $\text{CH}_2\text{OH}$



	R <sub>1</sub>	R <sub>2</sub>	R <sub>3</sub>	R <sub>4</sub>	
<b>1</b>	H	CHO	H	Me	Norstictic acid
<b>2</b>	Me	CHO	H	Me	Stictic acid
<b>3</b>	Me	CHO	H	$\text{CH}_2\text{OH}$	Constictic acid
<b>4</b>	Me	CHO	Me	Me	Methylstictic acid
<b>5</b>	Me	$\text{CH}_2\text{OH}$	H	Me	Cryptostictic acid
<b>6</b>	Me	Me	H	Me	Hypostictic acid

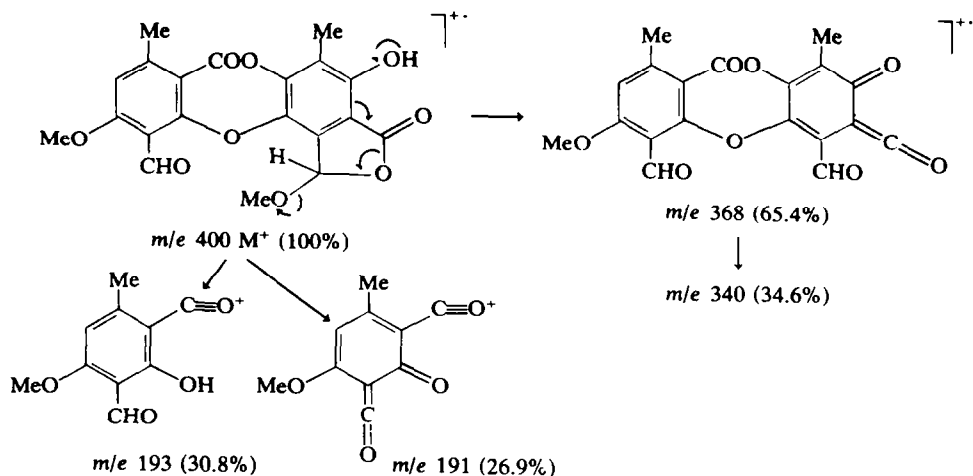
Methylstictic acid (**4**),  $\text{C}_{20}\text{H}_{16}\text{O}_9$ , colourless needles, mp  $250\sim 251^\circ$ , gave a purple colour with  $\text{FeCl}_3$  simi-

on the A ring was proved by the MS fragmentation giving a peak  $m/e$  177 (41.9%) (Scheme 2).

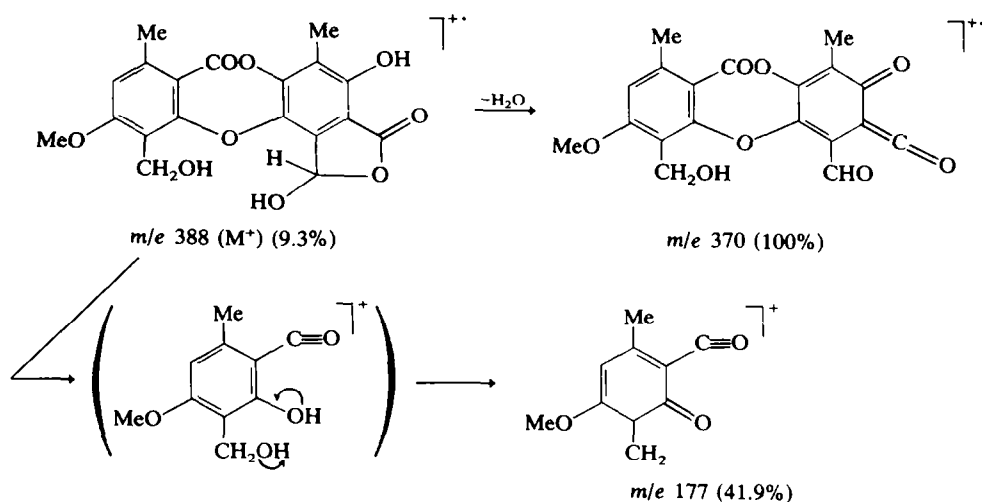
The structure of **5** has also been chemically proved by comparison with the established structure of **2**. On catalytic reduction of **2** with Pd-C in a mixture of EtOAc and HOAc (4:1) **5** was afforded, which was identified by mmp and comparisons of TLC and IR with the naturally isolated sample. Asahina *et al.* [1] reported that catalytic reduction of **2** with Pd-C in

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Scheme 1.



Scheme 2.

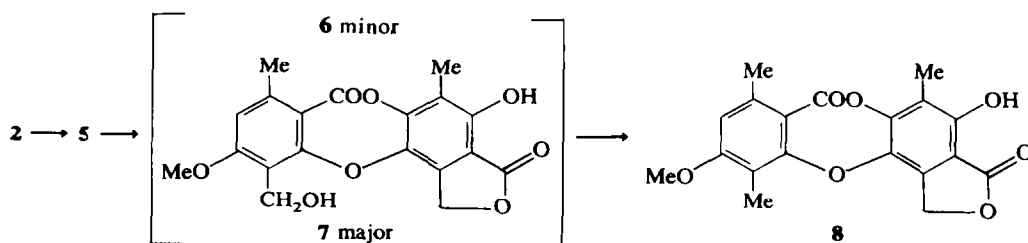
HOAc yielded hypostictic acid (**6**). In the present study **2** was reduced with  $\text{PtO}_2$  as catalyst in a mixture of EtOAc and HOAc (4:1) for a longer time to produce cryptostictinolide (**7**) as a major product and hypostictic acid (**6**) as a minor one in an earlier stage of reaction, which were finally converted into hypostictinolide (**8**) (Scheme 3).

Thus a solvent effect has been shown in the course of catalytic reduction of the CHO group of **2**.

#### EXPERIMENTAL

IR spectra were measured in KBr and  $^1\text{H}$  NMR spectra in  $\text{DMSO}-d_6$  with TMS as an int. standard at 100 MHz.

*Lobaria oregana* (Tuck.) Müll. Arg. This was collected in British Columbia, Canada in 1975. 6 kg were extracted  $\times 3$  with  $\text{C}_6\text{H}_6$  at room temp. to separate (+)-usnic acid (1.47%) and ergosterol peroxide (0.01%). Residue was extracted with  $\text{CHCl}_3$  and then  $\text{Me}_2\text{CO}$  at room temp. From the



Scheme 3.

supernatant of the conc Me<sub>2</sub>CO-extracts by 0.5 N oxalic acid-impregnated Si gel column chromatography using C<sub>6</sub>H<sub>6</sub>-Me<sub>2</sub>CO as solvent, norstictic acid (1), stictic acid (2) and constictic acid (3) were separated as the major products (total yield: ca 8%) and methylstictic acid (4) and cryptostictic acid (5) as the minor ones in a yield of 0.02 and 0.03%, respectively.

**Methylstictic acid (4).** C<sub>20</sub>H<sub>16</sub>O<sub>9</sub>, colourless needles (from Me<sub>2</sub>CO), mp 250–251° (decomp.), [α]<sub>D</sub><sup>20</sup> = ±0°; UV λ<sub>max</sub><sup>EtOH</sup> nm(log ε): 238 (4.62), 312 (3.77); IR ν<sub>max</sub><sup>KBr</sup> cm<sup>-1</sup>: 3370, 1755, 1740, 1695; <sup>1</sup>H NMR (DMSO-*d*<sub>6</sub>) δ ppm: 2.22 (3H, s, Me at C-3'), 2.48 (3H, s, Me at C-6), 3.44 (3H, s, MeO), 3.91 (3H, s, MeO at C-4), 6.44 (1H), 7.08 (1H, s, arom. H), 10.3 (1H, s, OH), 10.39 (1H, s, CHO); <sup>1</sup>H NMR (Py-*d*<sub>5</sub>) δ ppm: 2.36 (3H, s, Me at C-3'), 2.58 (3H, s, Me at C-6), 3.68 (3H, s, OMe), 3.81 (3H, s, OMe at C-4), 6.62 (1H), 6.85 (1H, s, arom. H), 10.98 (1H, s, CHO); MS: *m/e* 400 (M<sup>+</sup>, 100), 368 (M<sup>+</sup> - MeOH), 340 (368 - CO), 193, 191; High resolution MS: Observed M<sup>+</sup>: 400.0793. Calc. for C<sub>20</sub>H<sub>16</sub>O<sub>9</sub> M<sup>+</sup>: 400.0793. FeCl<sub>3</sub>: purple.

**Cryptostictic acid (5).** C<sub>19</sub>H<sub>16</sub>O<sub>9</sub>, colourless needles, mp 242–244° (decomp.) (from aq. Me<sub>2</sub>CO) UV λ<sub>max</sub><sup>EtOH</sup> nm (log ε): 214 (4.54), 267 (3.98), 316 (3.51); IR ν<sub>max</sub><sup>KBr</sup> cm<sup>-1</sup>: 3370, 3110, 1736; <sup>1</sup>H NMR (in DMSO-*d*<sub>6</sub>) δ ppm: 2.17 (3H, s, Me at C-3'), 2.42 (3H, s, Me at C-6), 3.84 (3H, s, OMe), 4.60 (1H, d, *J* = 10 Hz, —CH<sub>2</sub>—OH), 4.72 (1H, d, *J* = 10 Hz —CH<sub>2</sub>—OH), 6.92 (2H, arom-H and O—CH—OMe), 8.20 (1H, s, OH), 10.05 (1H, s, OH); <sup>1</sup>H NMR (Py-*d*<sub>5</sub>) δ ppm: 2.34 (3H, s, Me at C-3'), 2.6 (3H, s, Me at C-6), 3.7 (3H, s, MeO), 5.39 (2H, —CH<sub>2</sub>OH), 6.79 (1H, s, arom-H), 8.03 (1H); MS: *m/e* 388 (M<sup>+</sup>), 370 (M<sup>+</sup>—H<sub>2</sub>O (100)), 177. Anal. Calc. for C<sub>19</sub>H<sub>16</sub>O<sub>9</sub>: C, 58.76; H, 4.15 Found: C, 58.58; H, 4.10%; FeCl<sub>3</sub>: bluish-purple.

**Triacetate of 5.** C<sub>25</sub>H<sub>22</sub>O<sub>12</sub>, colourless needles, mp 270–271.5° (decomp.); <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ ppm: 2.07, 2.19, 2.27, 2.39, 2.50 (3H each, s (2Me, 3OAc)), 3.85 (3H, s, OMe), 4.79 (1H, d, *J* = 12 Hz), 5.52 (1H, d, *J* = 12 Hz), 6.64 (1H, s, arom.-H), 7.48 (1H, s); MS: *m/e* 514 (M<sup>+</sup>), 472 (100).

**Catalytic reduction of 5 (formation of cryptostictinolide (7)).** PtO<sub>2</sub> (20 mg) was suspended in EtOAc (6 ml) and saturated with H<sub>2</sub> (1 hr), then a soln of 5 (30 mg) in a mixture of EtOAc and HOAc (4:1) (10 ml) was added and shaken for

4 hr. The product was extracted into EtOAc and the extract chromatographed on a 0.5 N oxalic acid impregnated Si gel column to isolate 7. Cryptostictinolide (7), C<sub>19</sub>H<sub>16</sub>O<sub>8</sub>, colourless needles, mp 275–276° (decomp.) (From Me<sub>2</sub>CO). <sup>1</sup>H NMR (Py-*d*<sub>5</sub>) δ ppm: 2.3 (3H, s, Me at C-3'), 2.49 (3H, s, Me at C-6), 3.6 (3H, s, OMe), 4.91 (2H, —CH<sub>2</sub>—O—c) 5.84 (2H, —CH<sub>2</sub>—OH), 6.68 (1H, s, arom.-H); MS: *m/e* 372 (M<sup>+</sup>), 254 (100).

**Catalytic reduction of 2 (formation of 5).** Stictic acid (2) (30 mg) was dissolved in a mixture of EtOAc (20 ml) and HOAc (5 ml) and shaken with Pd-C (10%) (30 mg) under a H<sub>2</sub> atmosphere for 1.5 hr. The product was chromatographed on a 0.5 N oxalic acid impregnated Si gel column using C<sub>6</sub>H<sub>6</sub>-Me<sub>2</sub>CO as the solvent to separate a main product, mp 242°, which was proved to be identical with 5 by mmp and comparisons of IR and TLC. Stictic acid (2) (50 mg) was reduced with Pt<sub>2</sub>O in a mixture of EtOAc (6 ml) and HOAc (14 ml) for 8 hr. After removing catalyst, the reaction mixture was chromatographed on a 0.5 N oxalic acid impregnated Si gel column to separate hypostictinolide (8) and cryptostictinolide (7).

**Hypostictinolide (8).** C<sub>19</sub>H<sub>16</sub>O<sub>7</sub>, colourless needles, mp 277–278.5° (from Me<sub>2</sub>CO); <sup>1</sup>H NMR (DMSO-*d*<sub>6</sub>) δ ppm: 2.14 (6H, s, Me at C-3' and C-3'), 2.37 (3H, s, Me at C-6), 3.81 (3H, s, OMe), 5.44 (2H, s), 6.84 (1H, s, arom.-H), 9.77 (OH); MS: *m/e* 356 (M<sup>+</sup>) (100), 179.

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